1. S. M. Shostakovskii, A. A. Retinskii, and A. V. Bobrov, Izv. Akad. Nauk SSSR, Ser. Khim., No. 8, 1818 (1974).

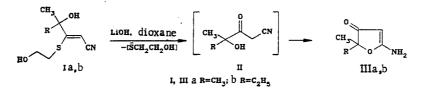
CYCLIZATION OF 3-HYDROXY-3-ALKYL-2-(2-HYDROXYETHYLTHIO)-1-

BUTENECARBONITRILES TO 5-AMINOFURAN-3-ONES

Yu. M. Skvortsov, E. I. Moshchevitina, UDC 547.722.6.07'339'269.1.04 and A. G. Mal'kina

3-Hydroxy-3-methyl-2-propylthio-1-butenecarbonitrile, obtained by the reaction between tertiary cyanoacetylenic alcohols and propanethiol, is highly stable, and does not undergo changes (isomerization or cyclization) either on keeping or on heating [1].

We have found that the sulfides (I), obtained from the same cyanoacetylenic alcohols and 2-mercaptoethanol, are very different in their chemical properties and undergo intramolecular cyclization to give, instead of the expected 1,3-oxathiolanes or iminodihydrofurans [2, 3], the 5-amino-2,2-dialkyl-2,3-dihydrofuran-3-ones (III).



An intermediate in the conversion of (I) into (III) appears to be the hydroxyketonitrile (II), formed by hydrolytic cleavage of the sulfide moiety. A similar cyclization has been observed when (I) is kept at room temperature (50 days).

<u>5-Amino-2,3-dihydro-2,2-dimethylfuran-3-one (IIIa).</u> A mixture of 0.3 g (1.6 mmole) of the sulfide (Ia), 0.03 g of LiOH, and 6 ml of dioxane was stirred for 3 h at 50°C. The dioxane was then removed under reduced pressure, and the residue dissolved in acetone and the solution filtered. Removal of the acetone gave 0.1 g (50%) of the aminofuranone (IIIa), mp 228-230°C (sublimes). ¹H NMR spectrum (DMSO-D₆), δ : 7.76 (2H, s), 4.28 (1H, s), 1.29 ppm (6H, s). ¹³C NMR spectrum (DMSO-D₆): 178.29 [C₍₅₎], 87.87 [C₍₄₎], 198.17 [C₍₃₎], 75.79 [C₍₁₎], 23.51 ppm (CH₃). IR spectrum (chloroform): 980...1050 (C-O-C), 1600 (C=CH), 1680 (C=O), 3420, 3450 cm⁻¹ (NH₂).

 $\frac{5-\text{Amino-2,3-dihydro-2-methyl-2-ethylfuran-3-one (IIIb)}{2} \text{ was obtained similarly in 68%} yield, mp 205-206°C (sublimes). PMR spectrum (DMSO-D₆), <math>\delta$: 7.80 (2H, s), 4.30 (1H, s), 1.50 (2H, q), 1.30 (1H, s), 0.99 ppm (3H, t). IR spectrum (chloroform): 980...1070 (C-O-C), 1600 (C=CH), 1680 (C=O), 3420, 3450 cm⁻¹ (NH₂).

LITERATURE CITED

- 1. A. N. Volkov, K. A. Volkova, E. P. Levanova, and B. A. Trofimov, Azv. Akad. Nauk SSSR, Ser. Khim., No. 1, 212 (1983).
- 2. A. S. Nakhmanovich, V. N. Elokhina, I. D. Kalikhman, and M. G. Voronkov, Khim. Geterotsikl. Soeding., No. 8, 1041 (1978).
- 3. Yu. M. Skvortsov, A. G. Mal'kina, B. A. Trofimov, A. N. Volkov, and V. M. Bzhezovskii, Zh. Org. Khim., <u>17</u>, 884 (1981).

Irkutsk Institute of Organic Chemistry, Siberian Branch, Academy of Sciences of the USSR, Irkutsk 664033. Translated from Khimiya Geterotsiklicheskikh Soedinenii, No. 9, pp. 1286-1287, September, 1988. Original article submitted November 24, 1987.